**8-sinf 1- variant Monitoring testi**

1. Elektramanfiyligi bir biridan keskin farq qiladigan modda qatorini tanlang?

 A) NaCl B) HCl C) NH3 D) PCl3

2. Qaysi elementlar erkin holda tura oladi? A) galogenlar B) ishqoriy C) inert D) amfoter

3. Fosfat kislotada (H 3PO 4) fosforning oksidlanish darajasi nechiga teng? A)+7 B)+1 C)+2 D)+5

4.Vodorodning oksidlanish darajasi -1 ga teng bo’lgan moddani aniqlang? A)NaH B)H 2CO 3 C)H2OD)H2O2

5.Quyidagi birikmalardan qaysi biri ion bog’lanishga ega? A)HF B)KF C)F2 D)CO2

6.Erkin holda elementlarning oksidlanish darajasi nechaga teng? A) -1 B) 0 C) -2 D) +2

7. Eritmalari va suyuqlanmalari elekrt tokini o’tkazuvchi moddalarga nima deyiladi? A) suyuqliklar B) noelektrolitlar C) elekrtolitlar D) to’g’ri javob yo’q

8. Olmos kristall tugunlari qanday shaklda joylasjgan? A) tetraedr B) kubsimon C) sakkiz burchak D) oval

9.Kichik davrlar qaysilar? A)1;2;3; B)5;6;4; D)1;5;9; C)4;5;6;

10. Necha xil bog’lanish turi bor? A) 3 ta B) 4 ta C) 5 ta D) 2 ta

11. Elektrmanfiylik deb nimaga aytiladi? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

12. Nima uchun olmos qattiq . Uning kristal panjara tugunlari qanday? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

13. S + O2 = SO2 quyidagi reaksiya tenglamani electron balans usulida tenglang?\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

14. Bariyning nisbiy atom massasi 137 ga teng, uning tartib raqami 56 ekanligini bilgan holda, bariy atom yarosidagi neytronlar sonini aniqlang. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

15. Metallmaslardan qaysi elementning oksidlanish darajasi o’zgarmas -1 ga teng?\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**8-sinf kimyo fanidan 1-variant kaliti**

|  |  |  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
| **№** | **1** | **2** | **3** | **4** | **5** | **6** | **7** | **8** | **9** | **10** |
| **Javoblar** | **A** | **C** | **D** | **A** | **B** | **B** | **C** | **A** | **A** | **B** |

11. Ayni element atomining boshqa element atomidan elektronlarni tortib olish xususiyati elektrmanfiylik deyiladi.

12. Atom kristall panjara. Hamma uglerod atomlari bir biriga juda qattiq tortilib turadi

13. S + O2 = SO2

14. 81ga teng

15. ftor

 **8-sinf 2- variant Monitoring testi**

1 .Ishqoriy metalldan boshlanib inert gazlar bilan tugaydigan elementlarning gorizontal qatoriga nima deyiladi ?

 A) guruh B) davr C) yonaki guruh D) asosiy guruh

 2. Katta davrlar qaysilar? A)1;2;3; B)5;6;4; D)1;5;9; C)4;5;6;

3. Kovolent bog’ necha turga bo’linadi? A)2 B)5 C)3 D)4

4. Kilorod qaysi qatordagi moddada -1 oksidlanish darajasini namoyon qiladi? A) Na2O B)CO2 C)P2O5 D)Na2O2

5.Quyidagi birikmalardan qaysi biri qutbsiz kovalent bog’lanishga ega? A)HF B)KF C)O2 D)CO2

6.Tipik metallarning tuzlari, ishqorlari va oksidlari qanday kristall panjaraga ega?

 A)metall B) molekulyar C)ionli D) atom

7. Donor akseptor bog’ga ega bo’lgan moddalarni toping?

 A) Suv B) kislota C)ammiak D)A va C javoblar to’g’ri

8. Elektrmanfiylik qiymati qaysi elementga nisbatan olingan?

A) Vodorod B) Litiy C) kaliy D) Ftor

9. Oltingugurt(S) ning oksidlanish darajasi berilgan qatorni toping?

 A)-2 B)+4 C) +6 D) barchasi to’g’ri

10. Kristal panjara tugunlari kubsimon joylashgan moddani belgilang?

 A) osh tuzi B) olmos C) grafit D) suv

11 . Oksidlanish darajasi deb nimaga aytiladi?\_\_\_\_\_\_\_\_\_

12 Metallarning oksidlanish darajasi nima uchun faqat musbat darajasini oladi?.\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

13. H 2CO3 kislotada uglerodning oksidlanish darajasi nechiga teng? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

14. Mg + O2 = MgO quyidagi reaksiya tenglamani electron balans usulida tenglang? \_\_\_\_\_\_\_\_\_\_\_\_

15 . Kristall panjara turlari necha xil? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

 **8-sinf kimyo fanidan 2-variant kaliti**

|  |  |  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
| **№** | **1** | **2** | **3** | **4** | **5** | **6** | **7** | **8** | **9** | **10** |
| **Javoblar** | **B** | **D**  | **A** | **D** | **C** | **C** | **D** | **B** | **D** | **D** |

11. Atomlarning olgan va bergan elektronlar soni shu atomning oksidlanish darajasi deyiladi?

12. Chunki barchasi electron bergani uchun

13. +4

14. 2Mg + O2 = 2MgO

15. 4 ga bo’linadi